

Saturation vapor pressure, e_{sat}

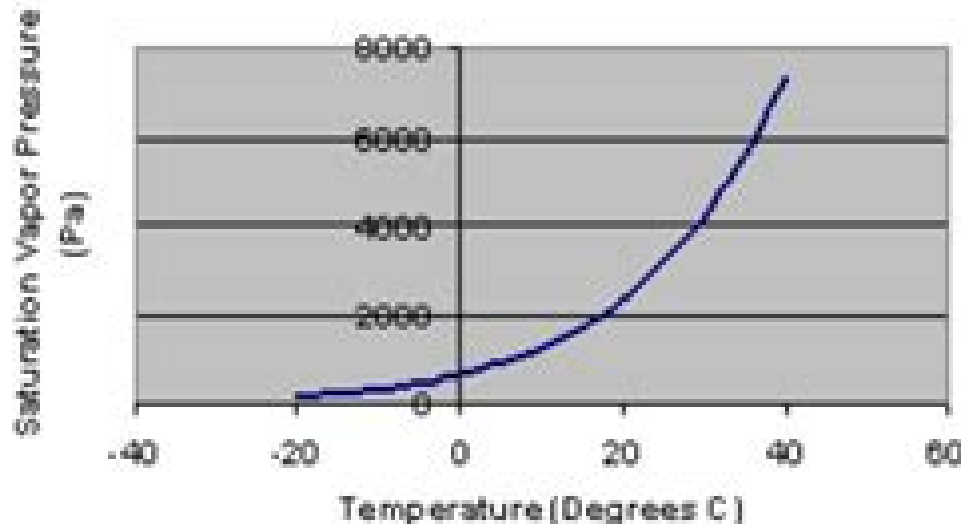
Saturation vapor pressure occurs when air is holding all the water vapor that it can at a given air temperature, then $RH = 100\%$ and $T = T_{dew\ point}$

Then $T_{dry\ bulb} = T_{wet\ bulb}$

$$e_{sat} = p_{ws} = \frac{e^{(77.345 + 0.0057 T - (7235 / T))}}{T^{8.2}}$$

<http://hurri.kean.edu/~yoh/calculations/atvap/satvap.html>

$$e_{sat} = p_{ws} = \frac{2.718^{(77.345 + 0.0057 T - (7235 / T))}}{T^{8.2}}$$



p_{ws} = water vapor saturation pressure (Pa)

e = the constant 2.718.....

T = dry bulb temperature of the moist air (K)

http://www.engineeringtoolbox.com/water-vapor-saturation-pressure-air-d_689.html

Saturation Vapor Pressure

Milo Milton Hasting



Saturation Vapor Pressure

Practical Meteorology Roland Stull,2018 A quantitative introduction to atmospheric science for students and professionals who want to understand and apply basic meteorological concepts but who are not ready for calculus

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On the Computation of Saturation Vapor Pressure Francis W. Murray,1966 Some problems involved in the computation of saturation vapor pressure of water for meteorological applications are discussed Two formulations for saturation vapor pressure in the pure phase over a plane surface of water or ice are considered namely the Goff Gratch formulation which is the internationally accepted standard and the Tetens formulation which is widely used because of its simplicity It is concluded that the Tetens formulation gives results sufficiently close to those of the Goff Gratch formulation for most meteorological applications Author

Reglement-tarief voor het bezigen der waterwerktuigen [van het] Koninklijk stapelhuis ,1899

New Calculation of Saturation Specific Humidity and Saturation Vapor Pressure in the COLA Atmospheric General Circulation Model Larry Marx,2002

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Saturation Vapor Pressure, Enthalpy and Entropy of Vaporization of Some Perfluorodi-n-alkyl Ethers R. M. Varushchenko,L. L. Pashchenko,1989

Vapor Pressure Determinations by the Saturation Method Drury Lee Weatherhead,1908

Vapor Pressure Determinations by the Saturation Method (Classic Reprint) Drury Lee Weatherhead,2018-03-19 Excerpt from Vapor Pressure Determinations by the Saturation Method The great exoerimental difficulties that attend the determination of the vapor pressure of solutions have pre hibited the collection of much data in this line The chief difficulties have been due to the lack of suitable apparatus several forms of which have been used Without very good set isfaction About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works

The Measurement of Vapor Pressure Lowering by the Air Saturation Method Edward Wight Washburn,Edward Otto Heuse,1915

The Vapor Pressure and Saturated Vapor and Liquid Densities of Isooctane (2,2,4-trimethylpentane) Fred Morgan Warzel,1948

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Measurement of Adsorption for Pressures Very Close to the Saturation Vapor Pressure [microform] Tanvir I. (Tanvir Iqbal) Farouk,2004 A two interface system of a single component fluid confined in a cylindrical container is a configuration in which a portion of a liquid phase would be above the vapor phase and a portion below Based on Gibbs model of interphase a thermodynamic analysis of a two interface configuration predicts that at equilibrium the contact angle at the upper three phase line is necessarily smaller than that at the lower three phase The analysis also predicts that at equilibrium the pressure profile is the same as it would have been if the vapor phase were not there A capillary system is considered that contains a single component fluid and is exposed to a gravitational field A portion of the liquid is above and a portion below the vapor Three sets of experiments are conducted with water by varying the vapor phase length The experimental investigation supports both the theoretical predictions The large contact angles at the lower three phase line indicate an adsorption process occurring near the interface Using the alpha isotherm relations together with the Young and Gibbs adsorption equation the contact angles for the experiments are predicted The predictions are found to be in close agreement with the measurements

ENVPREDRSCHFAC Technical Paper Environmental Prediction Research Facility (U.S.),1972

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Thermal Physics of the Atmosphere Maarten H. P. Ambaum,2010-05-24 Thermal Physics of the Atmosphere offers a concise and thorough introduction on how basic thermodynamics naturally leads on to advanced topics in atmospheric physics The book starts by covering the basics of thermodynamics and its applications in atmospheric science The later chapters describe major applications specific to more specialized areas of atmospheric physics including vertical structure and stability cloud formation and radiative processes The book concludes with a discussion of non equilibrium thermodynamics as applied to the atmosphere This book provides a thorough introduction and invaluable grounding for specialised literature on the subject Introduces a wide range of areas associated with atmospheric physics Starts from basic level thermal physics Ideally suited for readers with a general physics background Self assessment questions included for each chapter Supplementary website to accompany the book

Vapor Pressure of the Chemical Elements Andreĭ Nikolaevich Nesmeĭanov,1963

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